

SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR
 (AUTONOMOUS)

B.Tech I Year I Semester Supplementary Examinations February-2024
APPLIED CHEMISTRY
 (Common to ECE & EEE)

Time: 3 Hours**Max. Marks: 60**(Answer all Five Units $5 \times 12 = 60$ Marks)**UNIT-I**

- 1 a Define Electrode Potential.
 b Derive the Nernst equation for a single electrode potential and write its applications.

OR

- 2 a Explain the construction and working of lead-acid battery.
 b Describe the working and uses of hydrogen-oxygen fuel cell.

UNIT-II

- 3 Derive Schrodinger wave equation.

CO2 L3 12M

OR

- 4 a Explain π -molecular orbital of benzene with a neat sketch.
 b Differentiate bonding and anti-bonding molecular orbitals.

UNIT-III

- 5 a Explain the mechanism of free radical addition polymerization.
 b What is functionality of monomer?

CO3 L2 8M

CO3 L1 4M

OR

- 6 a What are conducting polymers? How are they classified?
 b Discuss the synthesis and applications of any one conducting polymer.

CO4 L1 4M

CO4 L2 8M

UNIT-IV

- 7 Describe the principle, instrumentation and applications of UV-visible spectroscopy with neat block diagram.

CO5 L2 12M

OR

- 8 Explain the principle, working and applications of Thin Layer Chromatography.

CO5 L2 12M

UNIT-V

- 9 a Explain the principle and applications of semiconductors.
 b Classify the semiconductors with examples.

CO6 L2 6M

CO6 L2 6M

OR

- 10 a Discuss the applications of supra-molecules in Catalysis and medical fields.

CO6 L2 8M

- b What is meant nanomaterial? How the nanomaterials classified?

CO6 L1 4M

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